PHYS261 Atomic Physics and Physical Optics

Lecture Thursday 18. September 2008

Topics:

Helium; The Triplet - Singlet Story Exchange Interaction final touch Helium; Perturbation Theory, Variational Theory

Comment:

After the lecture - preliminary version;

From last time: Functions of operators

$$e^{i\alpha\begin{pmatrix}0&1\\1&0\end{pmatrix}} = \overline{I}\cos\alpha + i\sin\alpha\begin{pmatrix}0&1\\1&0\end{pmatrix}$$
$$\begin{pmatrix}0&1\\1&0\end{pmatrix}\begin{pmatrix}0&1\\1&0\end{pmatrix} = \begin{pmatrix}1&0\\0&1\end{pmatrix} \quad \overline{I} = 1 = \begin{pmatrix}1&0\\0&1\end{pmatrix}$$

Pauli Matrices

$$e^{i\alpha\begin{pmatrix}0&1\\1&0\end{pmatrix}} = \overline{I}\cos\alpha + i\sin\alpha\begin{pmatrix}0&1\\1&0\end{pmatrix}$$
$$\begin{pmatrix}0&1\\1&0\end{pmatrix}\begin{pmatrix}0&1\\1&0\end{pmatrix} = \begin{pmatrix}1&0\\0&1\end{pmatrix} \qquad \overline{I} = 1 = \begin{pmatrix}1&0\\0&1\end{pmatrix}$$

The story you should learn to perform: (three slides follow in the ODP file)

The two electron atoms have the spin symmetry feature. You can make the total wavefunction antisymmetric by taking symmetric spin and space function antisymmetric. Or you can take antisymmetric spin and then this aplies space function is symmetric. Symmetric spins are called triplet, because there are 3 of them, antisymmetric is a singlet.

Two electrons repel each other. The repulsion adds energy, positive potential.

We will explain HOW to see that the repulsion is larger for singlets than for corresponding triplet.

Think e.g. about 1s 2p orbitals, as states a and b. Then schematically

Psi(antisym) -> a(1) b(2) - b(1) a(2)Psi(sym) -> a(1) b(2) + b(1) a(2)

when the repulsion is greatest? It is when $r1 \rightarrow r2$ - and then Psi(sym) is BIG, Psi(antisym) is very small ...

So HERE IS THE STORY:

triplet -> spin SYM -> space ASYM -> for r1->r2 Psi(antisym) is very small -> the repulsion is very small. **singlet** -> spin ASYM -> space SYM -> for r1->r2 Psi(sym) is BIG, the repulsion is BIG,

The story you should learn to perform:

The two electron atoms have the spin symmetry feature. You can make the total wavefunction antisymmetric by taking symmetric spin and space function antisymmetric. Or you can take antisymmetric spin and then this aplies space function is symmetric. Symmetric spins are called triplet, because there are 3 of them, antisymmetric is a singlet.

Two electrons repel each other. The repulsion adds energy, positive potential.

We will explain HOW to see that the repulsion is larger for singlets than for corresponding triplet.

The story you should learn to perform:

The two electron atoms have the spin symmetry feature. You can make the total wavefunction antisymmetric by taking symmetric spin and space function antisymmetric. Or you can take antisymmetric spin and then this aplies space function is symmetric. Symmetric spins are called triplet, because there are 3 of them, antisymmetric is a singlet.

Two electrons repel each other. The repulsion adds energy, positive potential.

We will explain HOW to see that the repulsion is larger for singlets than for corresponding triplet.

Think e.g. about 1s 2p orbitals, as states a and b. Then schematically

Psi(antisym) -> a(1) b(2) - b(1) a(2) Psi(sym) -> a(1) b(2) + b(1) a(2)

when the repulsion is greatest? It is when $r1 \rightarrow r2$ - and then Psi(sym) is BIG, Psi(antisym) is very small ...

The story you should learn to perform:

The two electron atoms have the spin symmetry feature. You can make the total wavefunction antisymmetric by taking symmetric spin and space function antisymmetric. Or you can take antisymmetric spin and then this aplies space function is symmetric. Symmetric spins are called triplet, because there are 3 of them, antisymmetric is a singlet.

Two electrons repel each other. The repulsion adds energy, positive potential.

We will explain HOW to see that the repulsion is larger for singlets than for corresponding triplet.

Think e.g. about 1s 2p orbitals, as states a and b. Then schematically

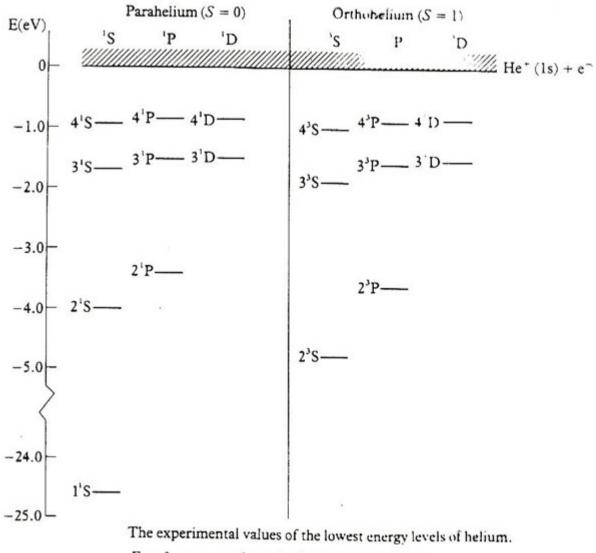
Psi(antisym) -> a(1) b(2) - b(1) a(2)Psi(sym) -> a(1) b(2) + b(1) a(2)

when the repulsion is greatest? It is when $r1 \rightarrow r2$ - and then Psi(sym) is BIG, Psi(antisym) is very small ...

So HERE IS THE STORY:

triplet -> spin SYM -> space ASYM -> for r1->r2 Psi(antisym) is very small -> the repulsion is very small. **singlet** -> spin ASYM -> space SYM -> for r1->r2 Psi(sym) is BIG, the repulsion is BIG,

Experiment - Level scheme for Helium



E = 0 corresponds to the ionisation threshold.

The Perturbation Theory (see next slide - work)

$$\left[-\frac{\hbar^2}{2m_e}\nabla_{r_1}^2 - \frac{Z\ e^2}{r_1} - \frac{\hbar^2}{2m_e}\nabla_{r_2}^2 - \frac{Z\ e^2}{r_2} + \frac{e^2}{|\mathbf{r}_1 - \mathbf{r}_2|}\right]\Psi\left(\mathbf{r}_1, \mathbf{r}_2\right) = E\Psi\left(\mathbf{r}_1, \mathbf{r}_2\right)$$

Repulsion expectation Value

Evaluation of the repulsion term using the multipole expansion

$$\frac{1}{|\mathbf{r}_{1} - \mathbf{r}_{2}|} = \sum_{LM} \frac{4\pi}{2L+1} \frac{r_{<}^{L}}{r_{>}^{L+1}} Y_{LM}^{\star}(\hat{r}_{1}) Y_{LM}(\hat{r}_{2})$$

where

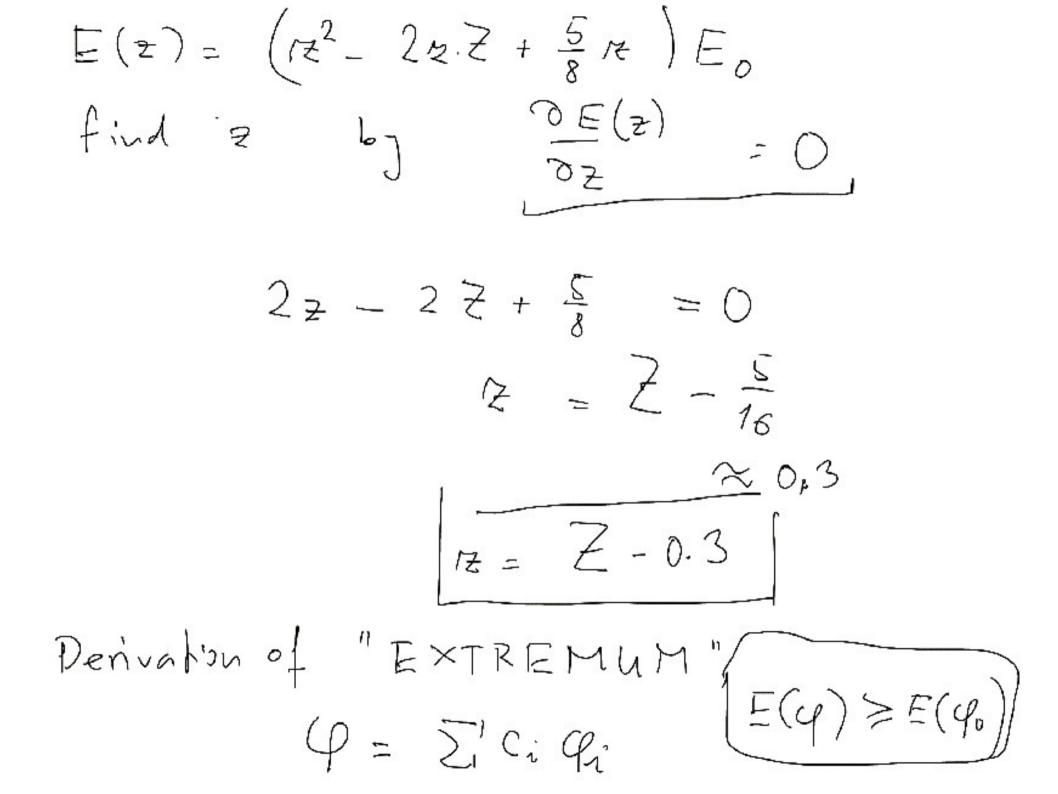
 $\int d^3 \mathbf{r}_1 \int d^3 \mathbf{r}_2 \; \psi_{100}^{\star}\left(\mathbf{r}_1
ight) \psi_{100}^{\star}\left(\mathbf{r}_2
ight) rac{e^2}{|\mathbf{r}_1 - \mathbf{r}_2|} \psi_{100}\left(\mathbf{r}_1
ight) \psi_{100}\left(\mathbf{r}_2
ight) = rac{5}{8} rac{Ze^2}{a_0}$

Perturbation theory result:

$$\mathbf{E}(\mathbf{z},\mathbf{u},\mathbf{u}) = -\frac{1}{2}\mathbf{z}^{2} - \frac{1}{2}\mathbf{z}^{2} + \frac{5}{8}\mathbf{z} \quad [\mathbf{a},\mathbf{u}]$$

The Variational Method

Perturbation theory $H = H_0 + H_1$ $H_0 \varphi = E_0 \varphi \left(\frac{H_0}{\varphi} = E_0 | \varphi \right)$ Hy is small (nonsense) $E = E_0 + \langle \varphi | H_1 | \overline{\varphi} \rangle E_1$ $= \langle \varphi | H | \varphi \rangle = \langle \varphi | H_0 | \varphi \rangle$ (first order....) + < 94419 [Repeat Pett. Th 6r read in Wikipo] $E_{0} = E_{elechold} \begin{pmatrix} z \\ 1 \end{pmatrix} + E_{elechold} \begin{pmatrix} z \\ n_{2} \end{pmatrix} \\ \frac{z}{n_{2}} \end{pmatrix} + \frac{1}{2} \frac{z^{2}}{n_{2}} + \begin{pmatrix} -\frac{1}{2} \frac{z^{2}}{n_{2}} \end{pmatrix} \\ \frac{z}{n_{2}} + \frac{1}{2} \frac{z^{2}}{n_{2}} \end{pmatrix} = \frac{5}{8} z$



To remember in preliminary version

```
Carousel (pictures t include??)
Flogiston - Look it up
```

The joke about Who wrote Hamlet - for remembering Stern-Gerlach (see wikipedia)

Quantiki (Quantum wiki) - very bad QM intro Quantum Computation