

PHYS261 Atomic Physics and Physical Optics

Lecture

Tuesday 7. October 2008

Topics:

Helium; Final touch

Helium; Excited States, Doubly excited states

Autoionizing states - also Auger Effect

Comment:

Revised version;

The Perturbation Theory (see next slide - work)

$$\left[-\frac{\hbar^2}{2m_e} \nabla_{\mathbf{r}_1}^2 - \frac{Z e^2}{r_1} - \frac{\hbar^2}{2m_e} \nabla_{\mathbf{r}_2}^2 - \frac{Z e^2}{r_2} + \frac{e^2}{|\mathbf{r}_1 - \mathbf{r}_2|} \right] \Psi(\mathbf{r}_1, \mathbf{r}_2) = E \Psi(\mathbf{r}_1, \mathbf{r}_2)$$

Repulsion expectation Value

Evaluation of the repulsion term using the multipole expansion

$$\frac{1}{|\mathbf{r}_1 - \mathbf{r}_2|} = \sum_{LM} \frac{4\pi}{2L+1} \frac{r_{<}^L}{r_{>}^{L+1}} Y_{LM}^*(\hat{r}_1) Y_{LM}(\hat{r}_2)$$

where

$$r_{<} = r_1, \quad r_{>} = r_2 \quad \text{for } |\mathbf{r}_1| < |\mathbf{r}_2|$$

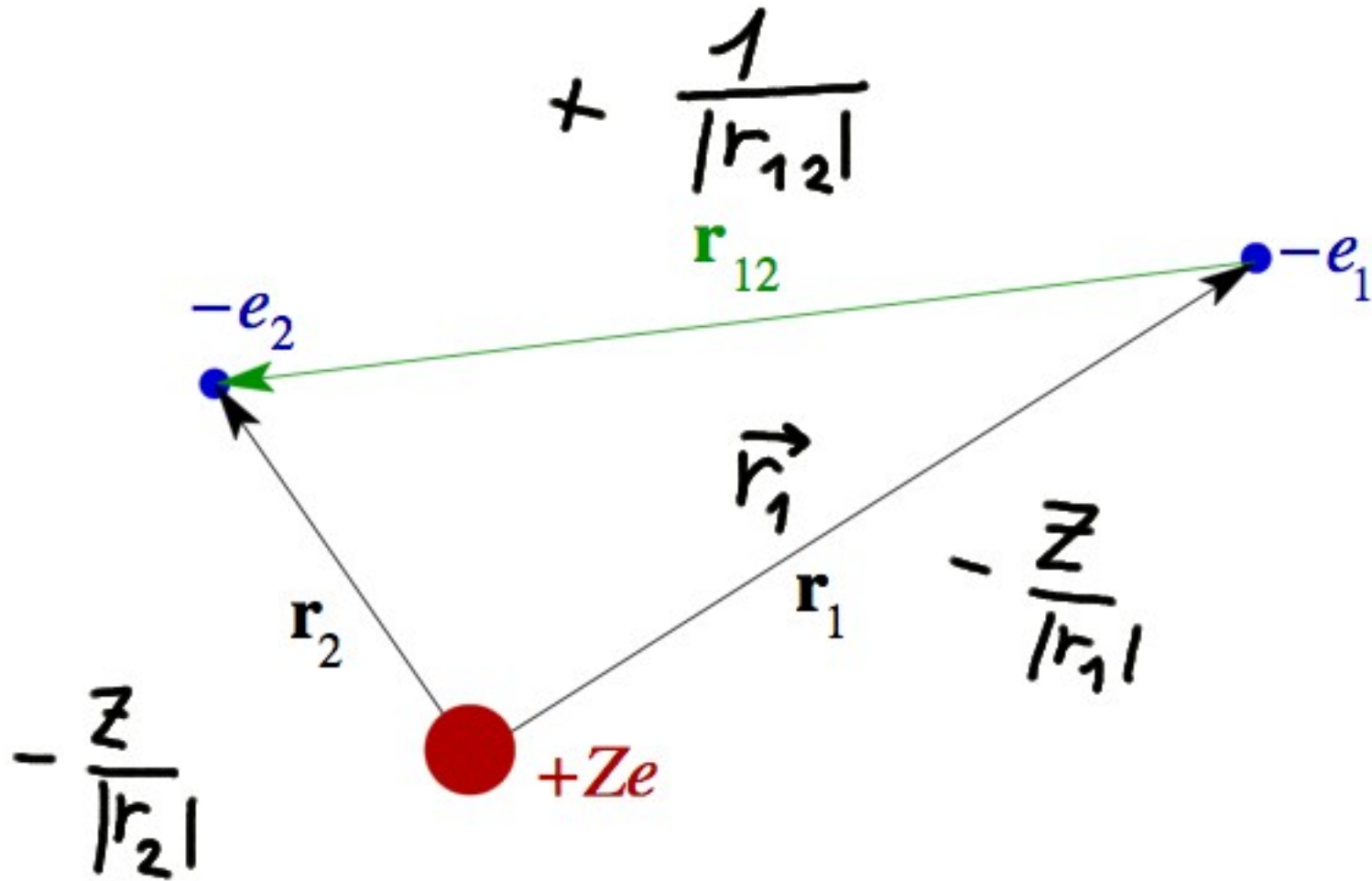
$$r_{<} = r_2, \quad r_{>} = r_1 \quad \text{for } |\mathbf{r}_1| > |\mathbf{r}_2|$$

Perturbation theory result:

$$\int d^3\mathbf{r}_1 \int d^3\mathbf{r}_2 \psi_{100}^*(\mathbf{r}_1) \psi_{100}^*(\mathbf{r}_2) \frac{e^2}{|\mathbf{r}_1 - \mathbf{r}_2|} \psi_{100}(\mathbf{r}_1) \psi_{100}(\mathbf{r}_2) = \frac{5}{8} \frac{Z e^2}{a_0}$$

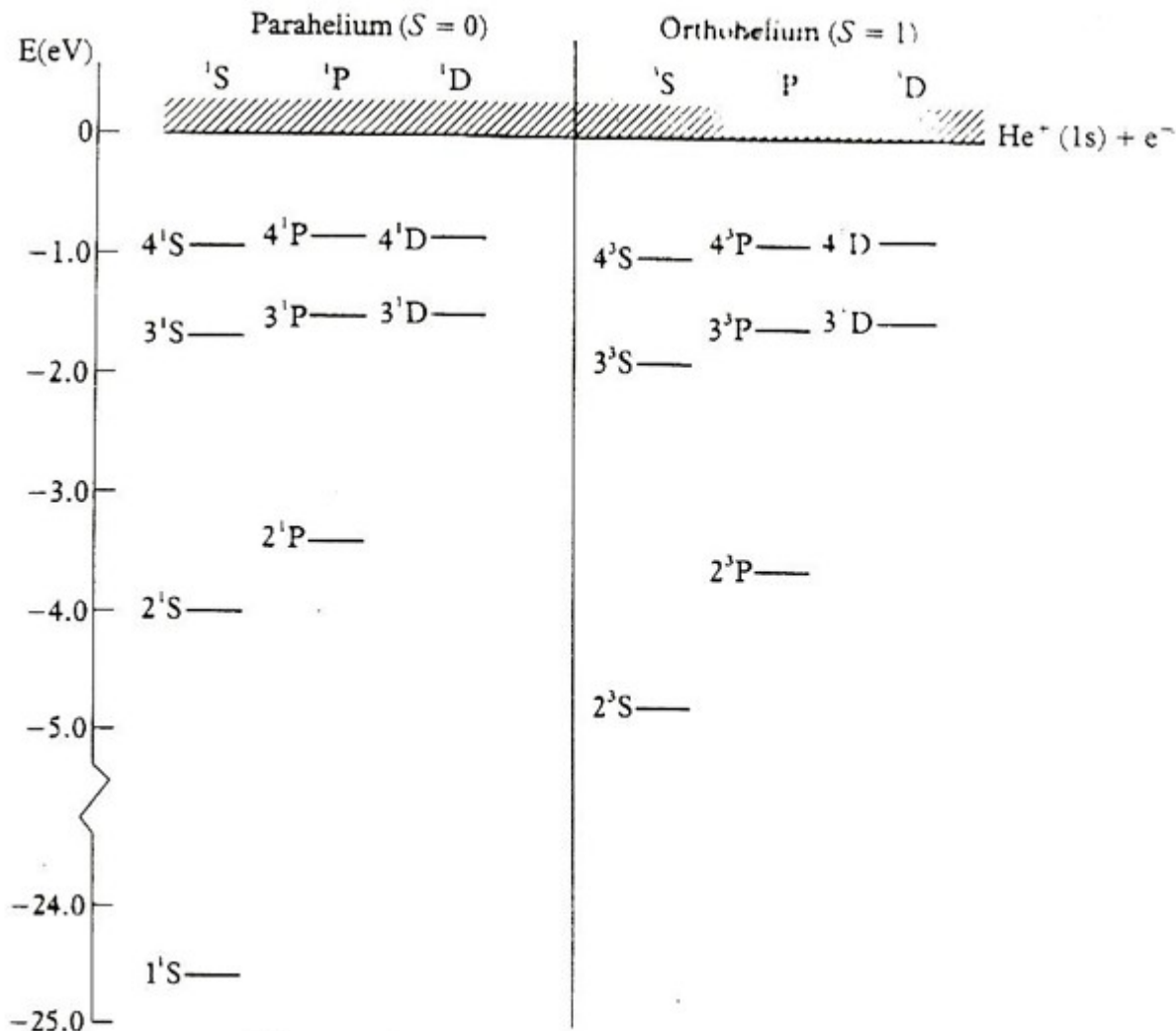
$$E(Z, 1s, 1s) = -\frac{1}{2} Z^2 - \frac{1}{2} Z^2 + \frac{5}{8} Z \quad [a.u.]$$

The Variational Method



H minus is also possible
 Negative ion of
 hydrogen

Experiment - Level scheme for Helium



The experimental values of the lowest energy levels of helium.

$E = 0$ corresponds to the ionisation threshold.

Variational Method basics

$$H\psi = E\psi \quad \psi(\vec{x}, \alpha) \quad \alpha - \text{parameter}$$

$$e^{-\alpha r} \cdot N(\alpha)$$

"best possible α "

i.e. gives best approximation to E ?

ENERGY E

$$\langle \psi(\alpha) | H | \psi(\alpha) \rangle \approx E(\alpha)$$

for which α $E(\alpha)$ closest to E .

$$\psi(\vec{x}, \alpha) = \sum_i c_i \varphi_i(\vec{x}) \quad | \quad H\varphi_i = E_i \varphi_i$$

$$\sum_i c_i^* \sum_j c_j \int \varphi_i^*(x) H(x) \varphi_j(x) d\tau$$

E_0 (or E_{1st} in H)

$$\langle E \rangle = \langle H \rangle = \sum_i |c_i|^2 E_i \geq \sum_i |c_i|^2 E_0$$

$$E_\alpha \rightarrow \langle E \rangle = \langle H \rangle \geq E_0$$

$$\langle \psi(\alpha) | \psi(\alpha) \rangle = \sum_i c_i^* \sum_j c_j \underbrace{\langle \varphi_i | \varphi_j \rangle}_{\delta_{ij}}$$

$$= \sum_i |c_i|^2 = 1$$

normalized

$$E(z) = \left(z^2 - 2z + \frac{5}{8} \right) E_0$$

find z by $\frac{\partial E(z)}{\partial z} = 0$

$$2z - 2z + \frac{5}{8} = 0$$

$$z = z - \frac{5}{16}$$

$$z = z - 0.3$$

Derivation of "EXTREMUM"

$$\varphi = \sum_i c_i \varphi_i$$

$$E(\varphi) \geq E(\varphi_0)$$

To remember in preliminary version

Carousel (pictures t include??)

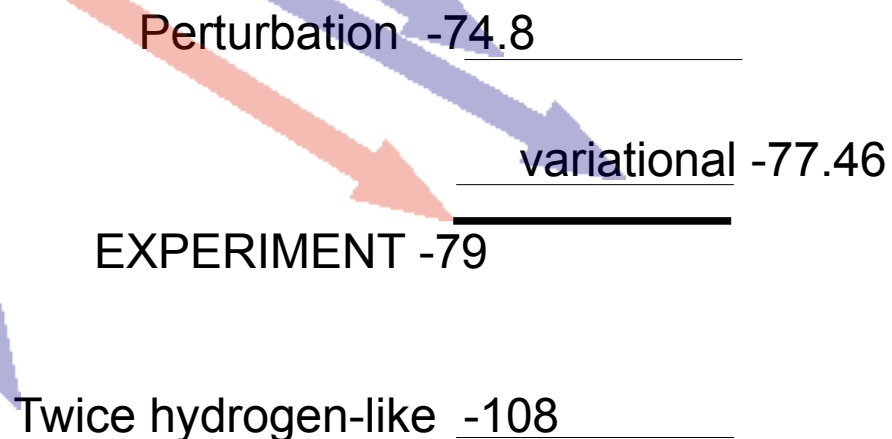
Flogiston - Look it up

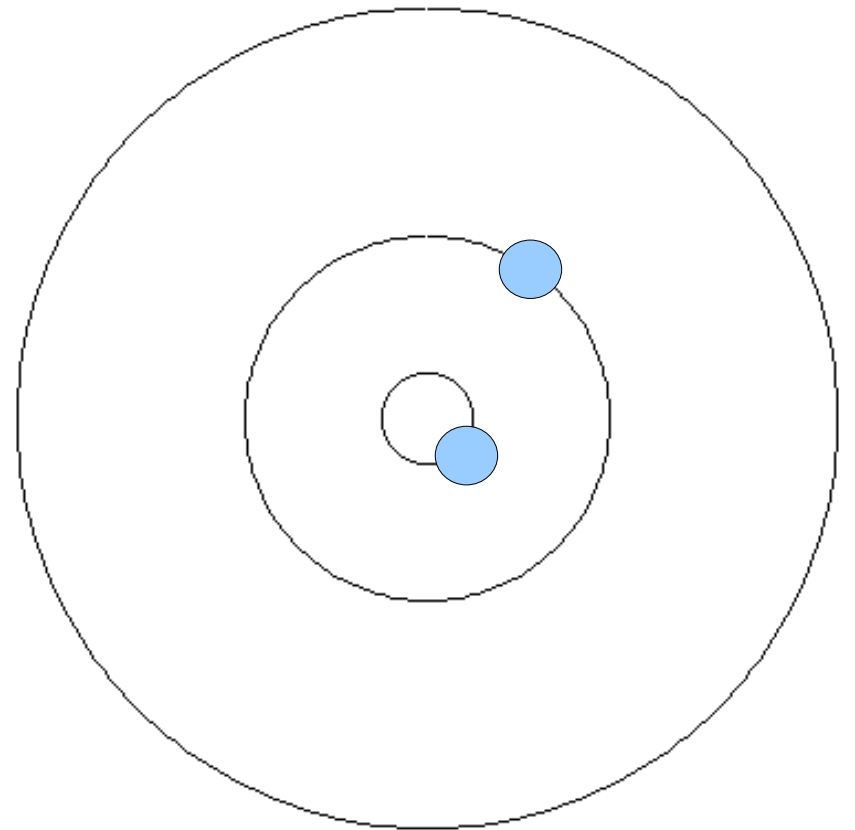
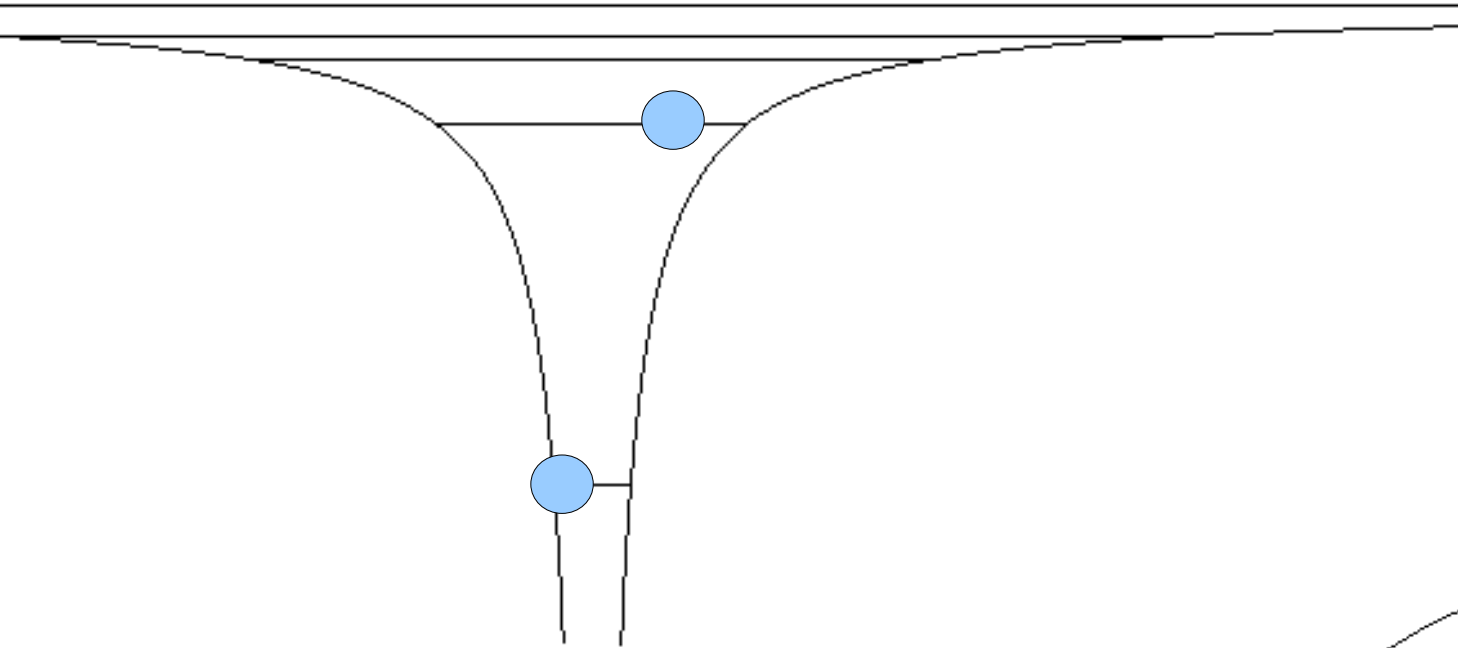
The joke about Who wrote Hamlet - for remembering
Stern-Gerlach (see wikipedia)

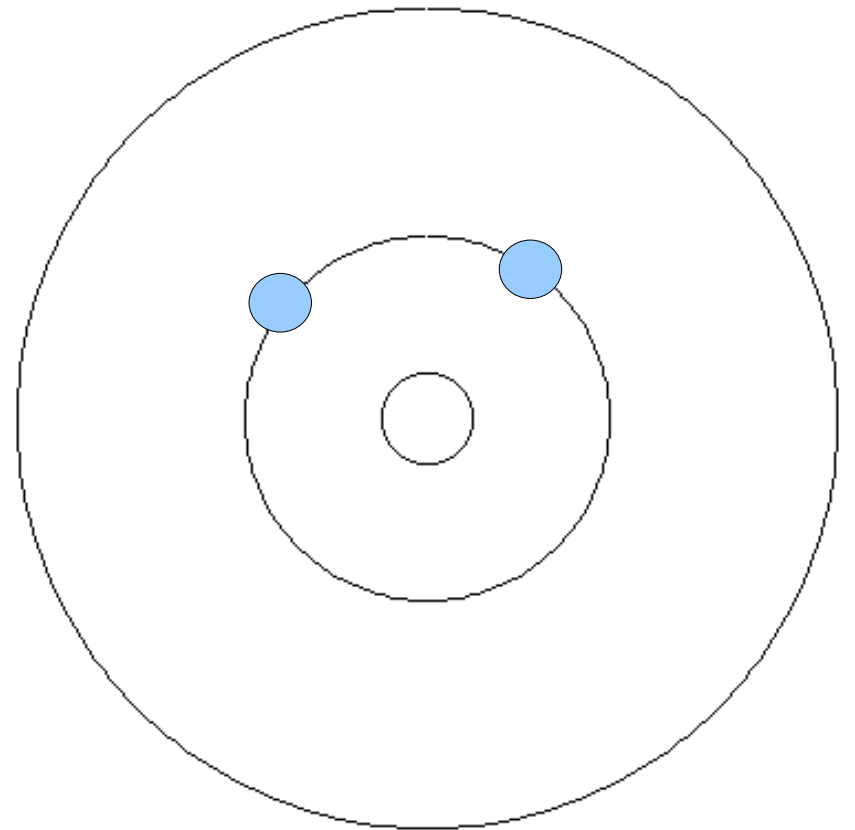
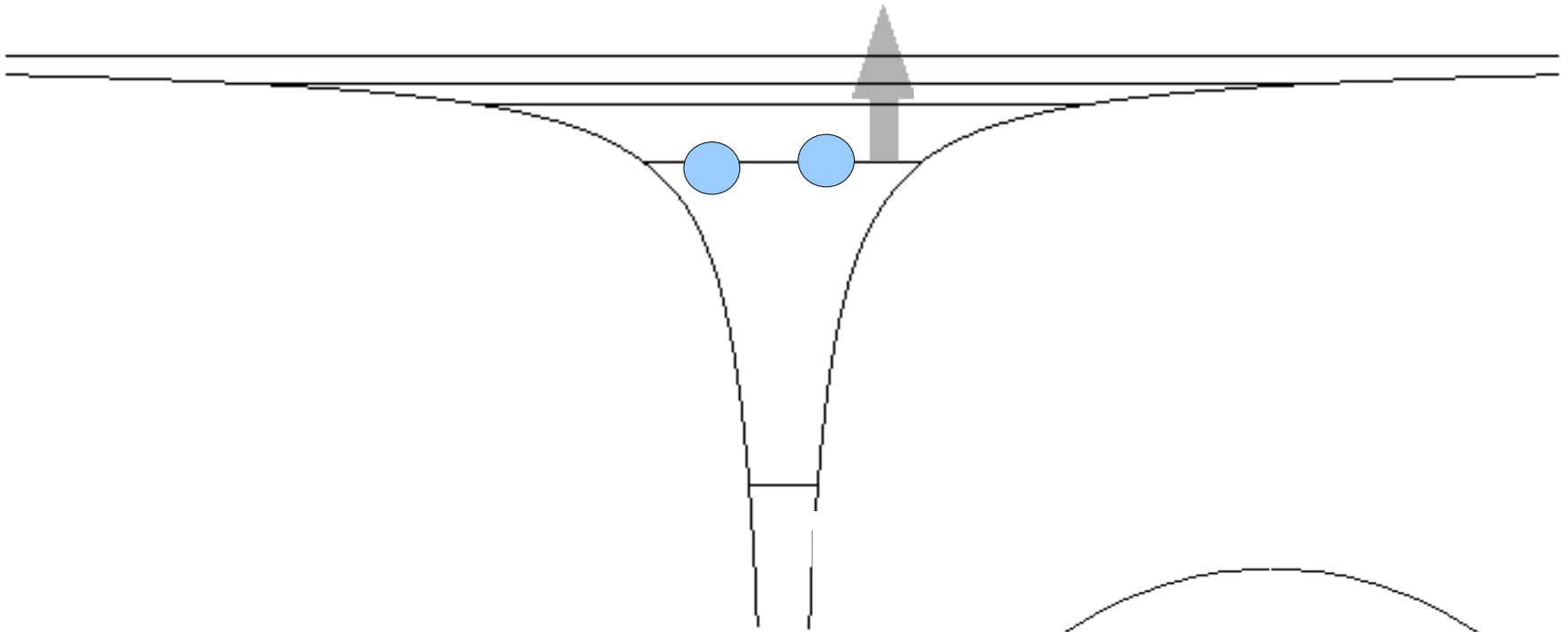
Quantiki (Quantum wiki) - very bad QM intro
Quantum Computation

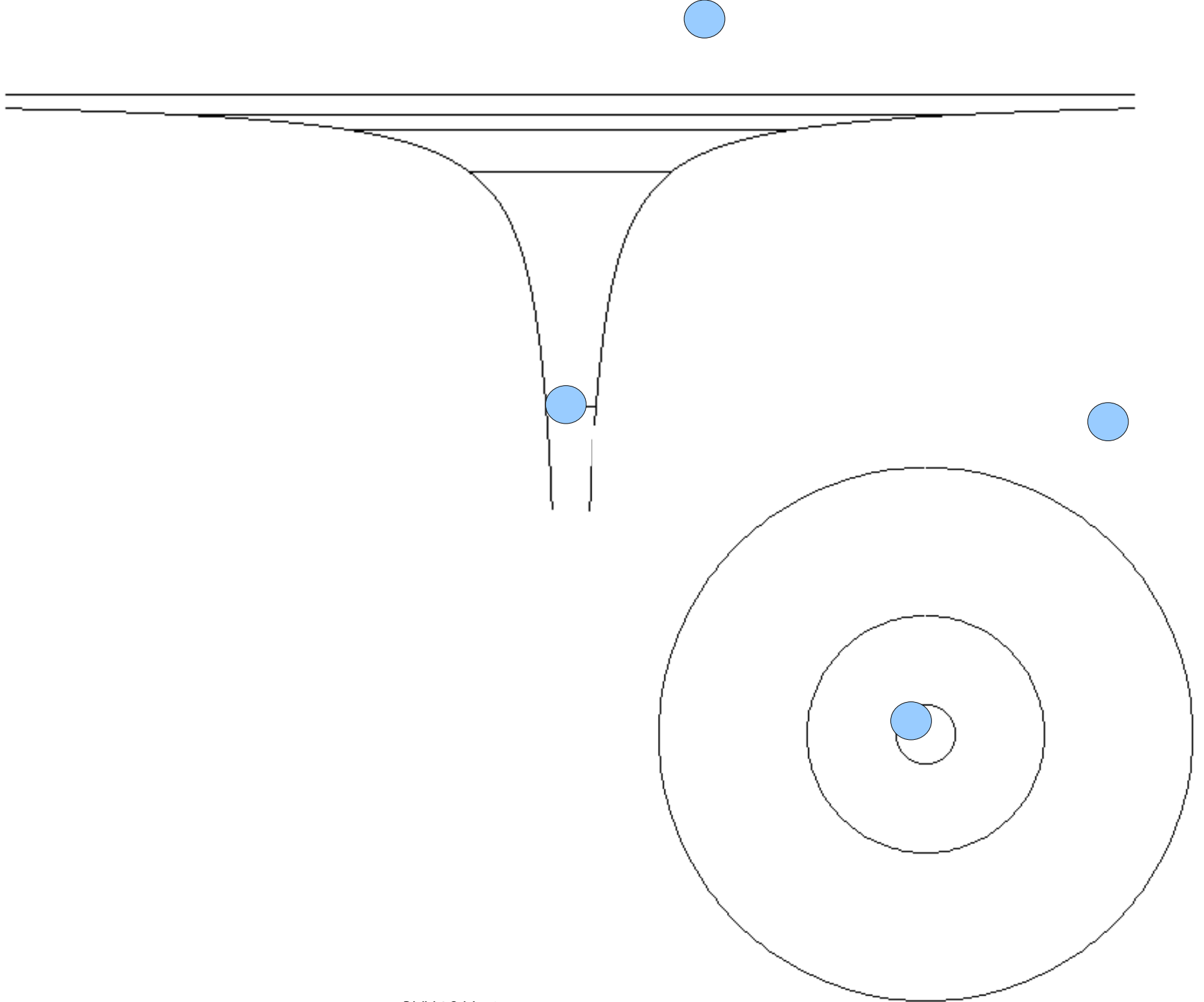
	H	He	Li ⁺	Be ⁺⁺	B ⁽³⁺⁾	C ⁽⁴⁺⁾
Z	1	2	3	4	5	6
Ion.pot.[au]	0,757	24,60	75,62	153,95	259,49	392,22
2.Ion.pot	13,600	54,40	122,40	217,60	340,00	489,60
EXP.BindEner	-14,357	-79,00	-198,02	-371,55	-599,49	-881,82
2 E _{1s}	-27,200	-108,80	-244,80	-435,20	-680,00	-979,20
2 E _{1s} + 5/8 Z	-10,200	-74,80	-193,80	-367,20	-595,00	-877,20
Variational	-12,856	-77,46	-196,46	-369,86	-597,66	-879,86
EXP.BindEner	-14,357	-79,00	-198,02	-371,55	-599,49	-881,82

- Ion.pot. Ionization potential: The energy to remove the first electron
- 2.Ion.pot Second Ionization potential: The energy to remove the second electron
- EXP.BindEner The experimental binding energy is sum of the two ionization potentials









Variational methods

effective $\propto e^{-\alpha r}$

More parameters one sees α_1, α_2 Antisym

$$\psi_{\alpha}(r_1, r_2) = \varphi_{\alpha}(r_1) \varphi_{\alpha}(r_2) \text{ independent}$$

IMPROVEMENTS: not independent

$$\psi(r_1, r_2) \propto \varphi_{\alpha}(r_1) \varphi_{\alpha}(r_2) f(r_{12})$$

$$\psi(r_1, r_2) \propto \varphi_{\alpha}(r_1) \varphi_{\alpha}(r_2) f'(r_{12})$$

- More than 1 parameter
- more terms

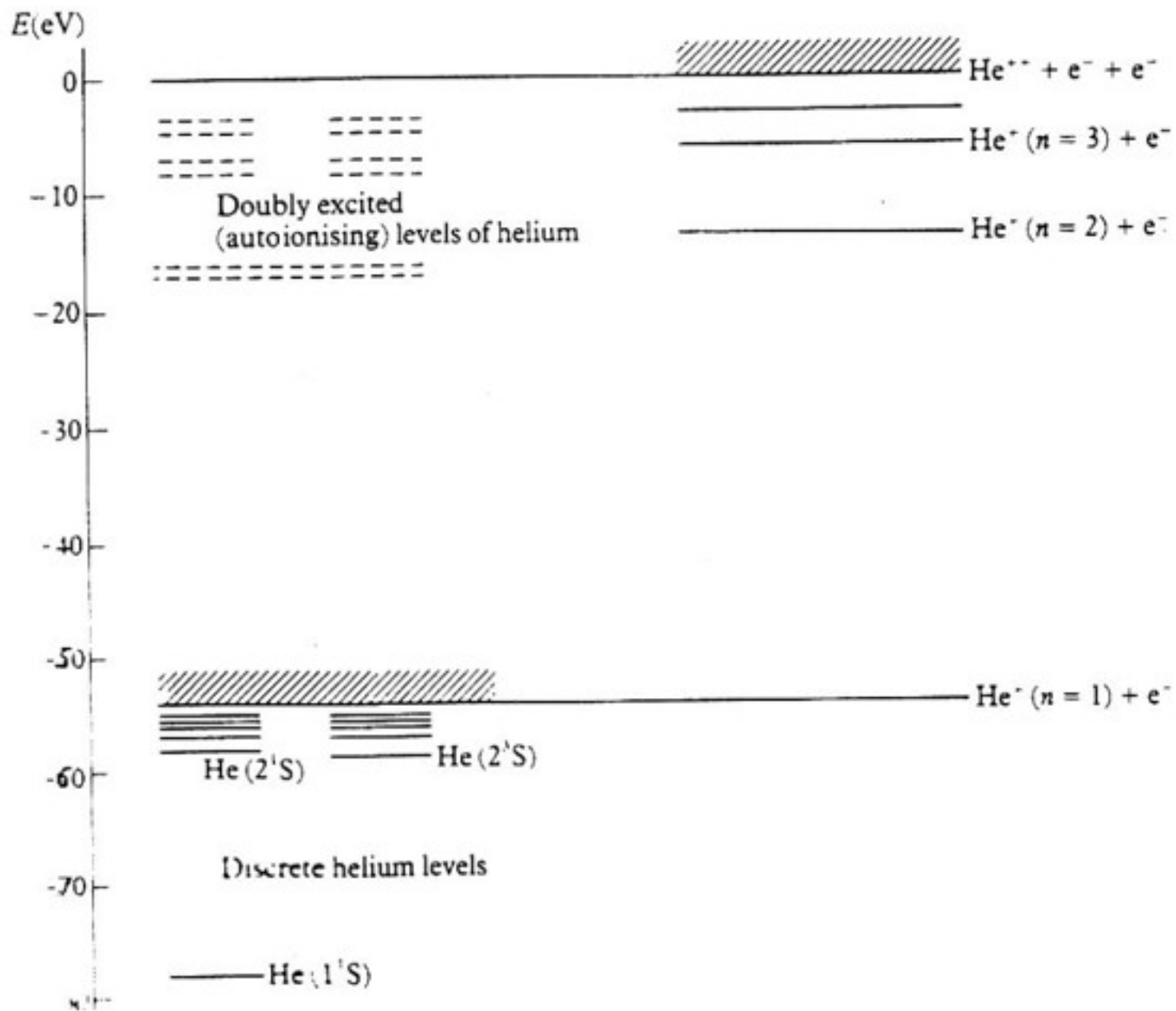
Hylleraas wavefunctions 1935

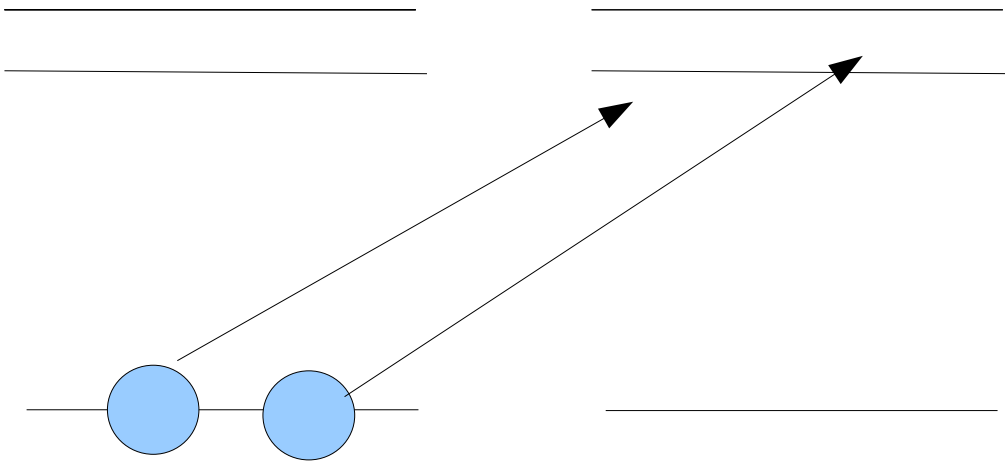
IN BERGEN

Most modern wavefunctions

over 1000 parameters

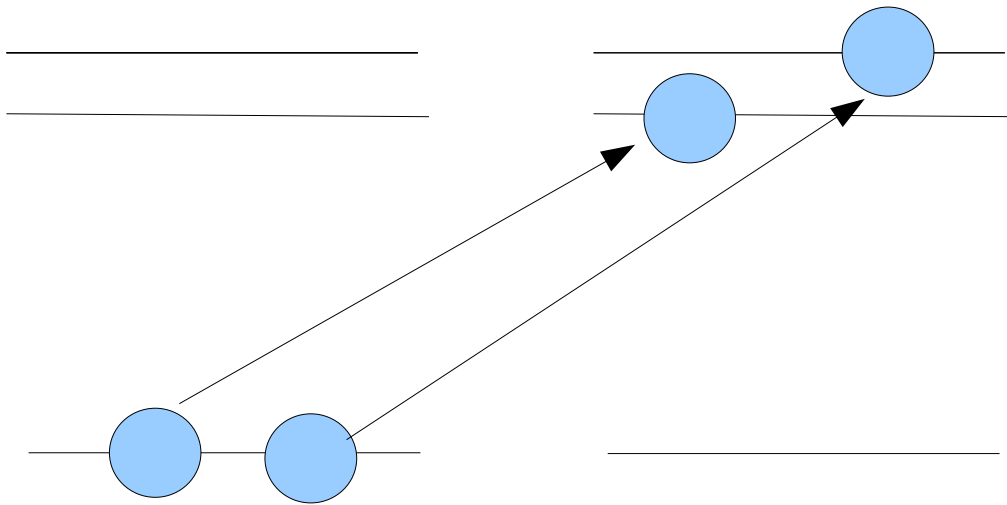
18 digits agreement with





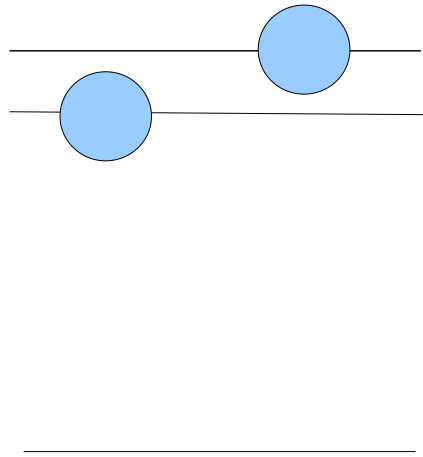
Autoionization
Autoionizing states

Auger Effect (autoionization from already ionized atom)



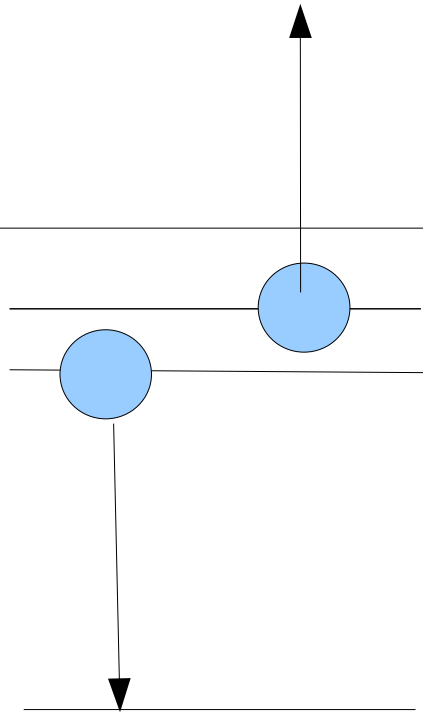
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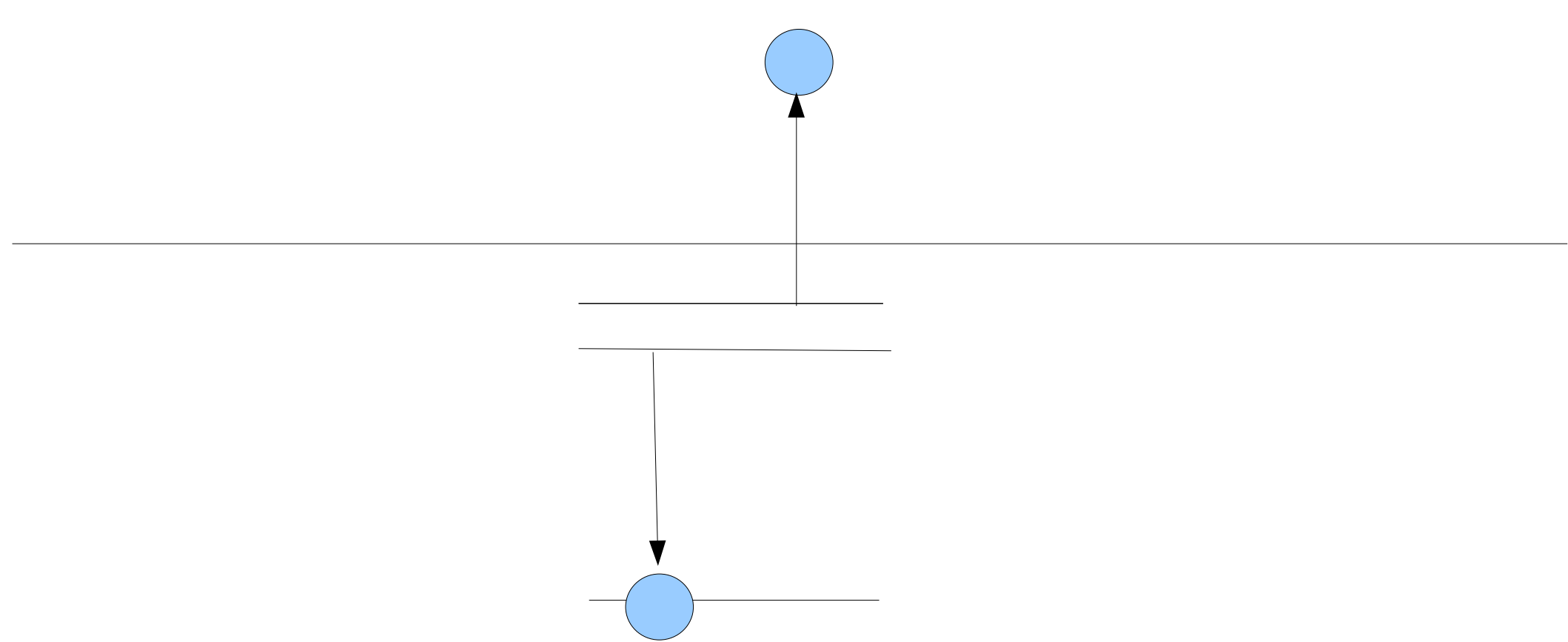
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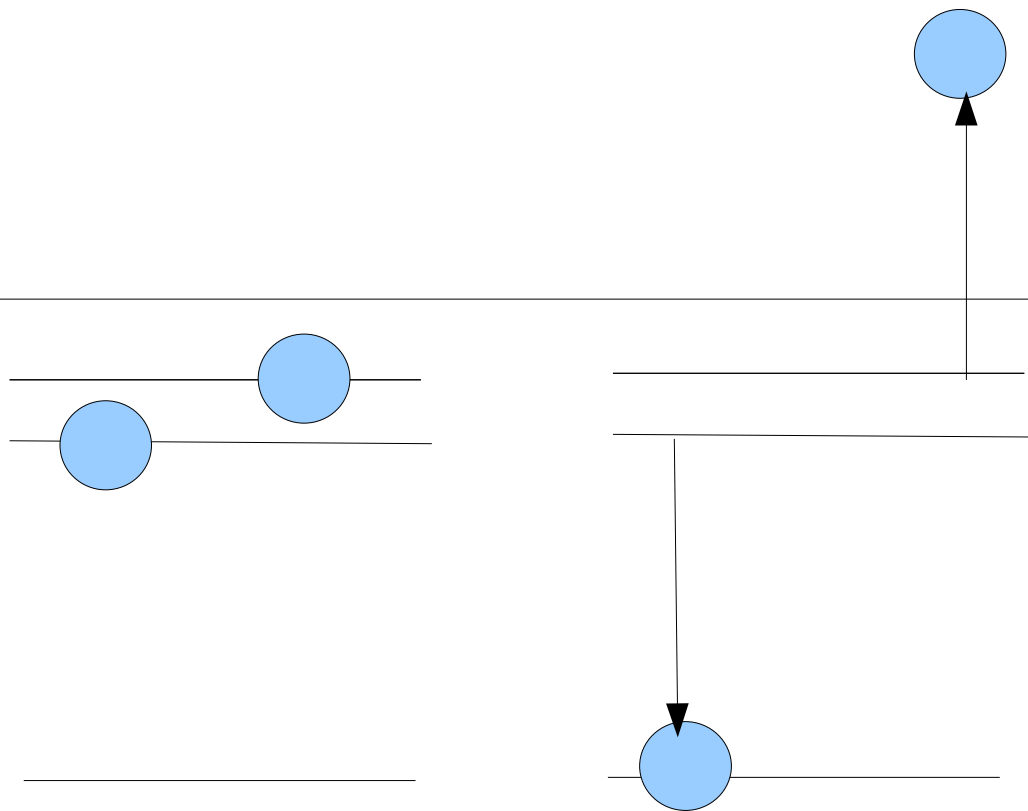
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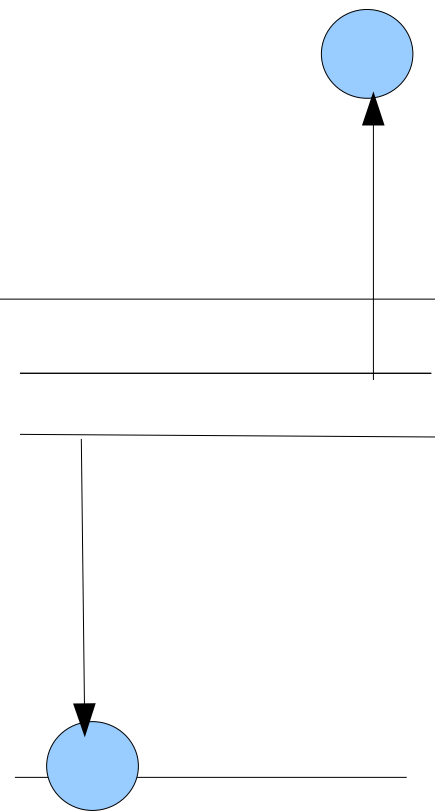
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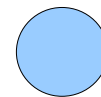
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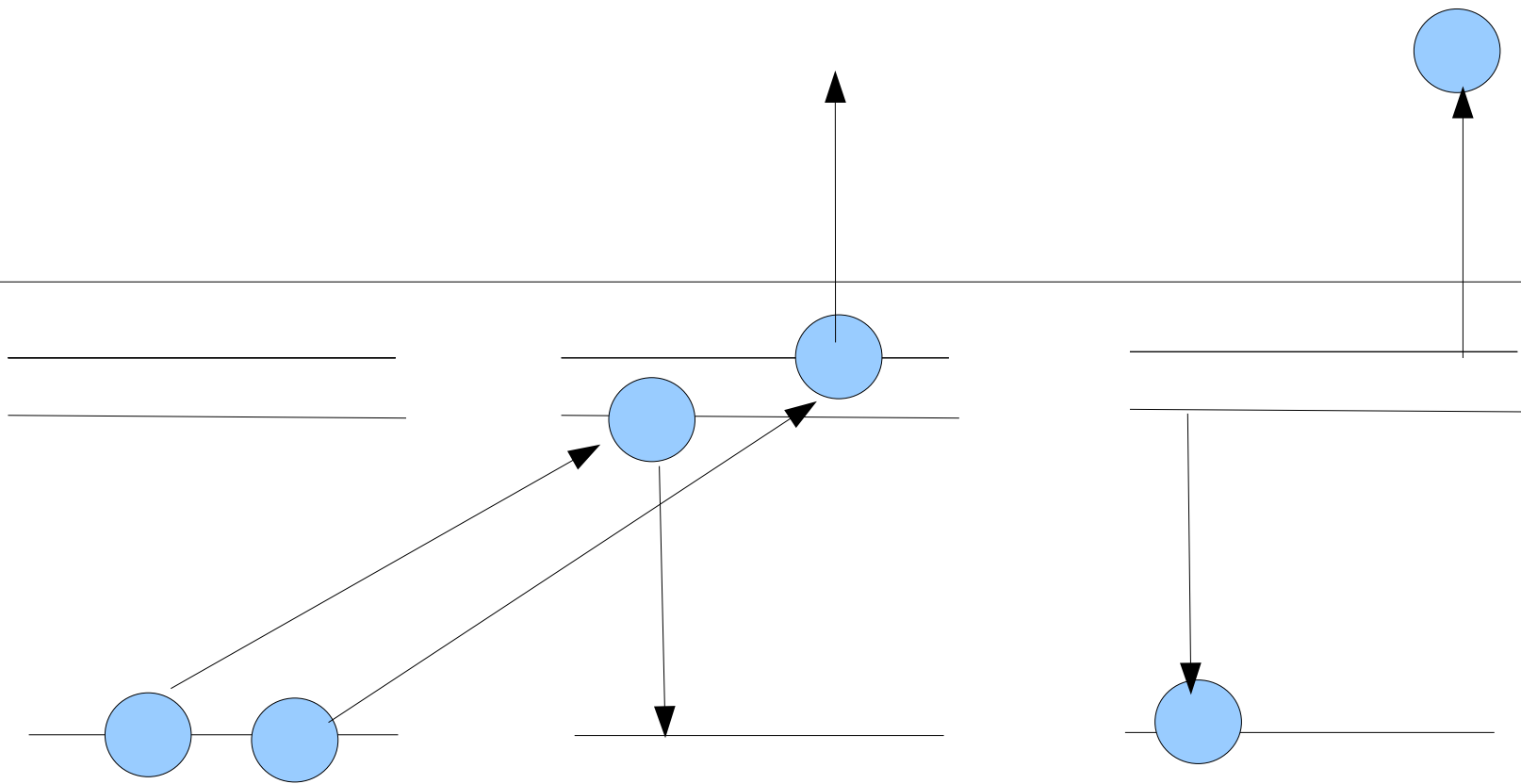
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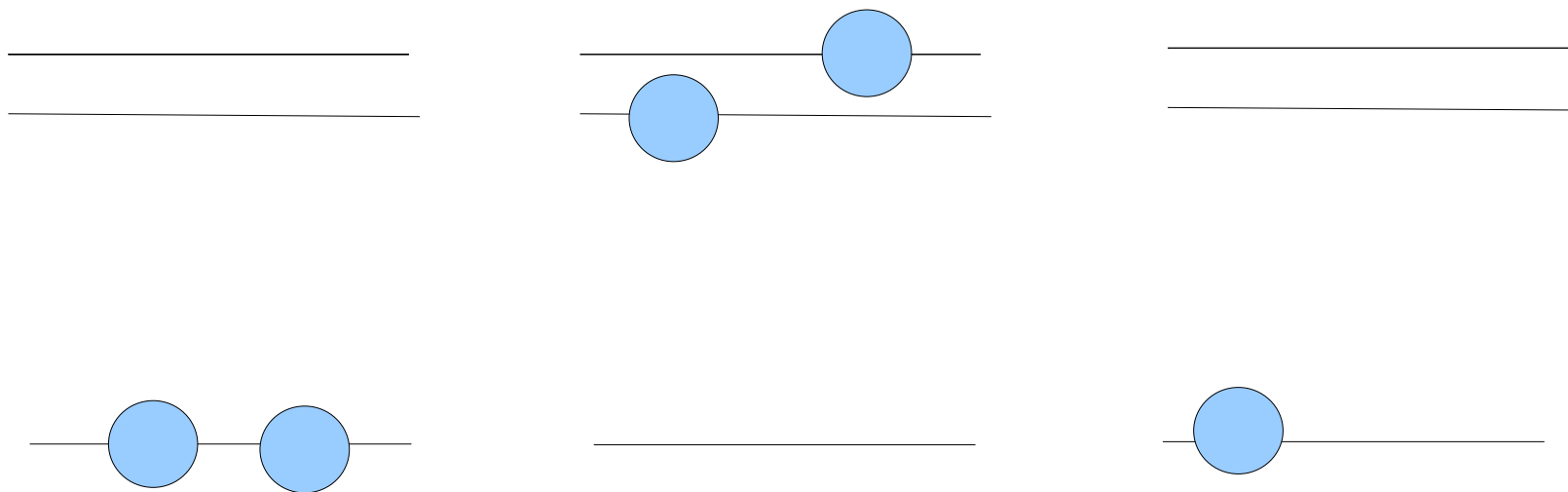
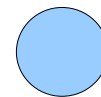
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Autoionization
Autoionizing states

Auger Effect (autoionization from already ionized atom)

Helium

Triplet \times Singlet (Spin)

Symmetric - Antisymmetric

—————→ Ferromagnetism

Perturbation → effective Z
→ variation

~~Independent electrons~~

Correlation → Hylleraas

Singly - excited

Doubly - excited → Anion

Many electron-atoms