## This note was written in "realtime" by Ingjald Pilskog. Edited by L. K. $Third\ lecture\ 29.08.06$

Atomic world construction  $a_0$  - radius, length  $\Delta x \approx a_0 \ \Delta k \approx 1/a_0$  $(\hbar \Delta k = \Delta p \text{ wavenumber})$  $\Delta E \approx \frac{(\Delta p)^2}{2m} = \frac{\hbar^2}{2ma_0^2} \approx \frac{\hbar^2}{ma_0^2} \text{ (kinetic energy T_0. ) Potential energy V_0 |} - \frac{e^2}{a_0}$ Atomic unit of length  $a_0 = 0,529$  Å=  $\frac{\hbar^2}{e^2 m}$  Lenght unit Atomic unit of energy Energy unit  $\frac{e^2}{a_0} = \frac{\hbar}{ma_0^2} = 27, 2 \text{ eV}$  $\langle T_0 \rangle = \langle V_0 \rangle$  should be of the same order. Atomic unit of velocity  $p_0 = \hbar k_0 = \frac{\hbar}{a_0}$ Atomic unit of time  $t_0 = \frac{a_0}{v_0} = (\frac{a_0}{e^2})\hbar = \frac{\hbar}{E_0}$ Alternative postulate  $t_0 = \frac{\hbar}{E_0}$ (Statement a.u.  $\longleftrightarrow e = m\hbar = 1$  is useless)  $\frac{0.66 \cdot 10^{-15}}{27.2}$ s=0,24 · 10<sup>-15</sup>  $\dot{\hbar}=0,66\cdot 10^{-15}~{\rm eVs}~t_0=$  $(2 \pi \text{ for angular freq.})$  $\nu$  frequency  $\omega$  - angular frequency  $k_0T$  is the "physical temperature". Room temperature is thus  $\frac{1}{40}$  eV or 25 meV Atomic unit of energy  $\longrightarrow$  VERY HOT

## More about bound states in H



States characterized by n, l (*m* - magnetic)

In H energies given by  $\frac{1}{n^2}(-\frac{1}{2})a.u.$ 

"Ladi says its nonsens to talk about m as a quantum number", wrote Ingjald. He did not write: "unless we have magnetic field on".

Angular momentum

$$L = \omega \mathcal{I}$$
  $T_{rot} = \frac{1}{2} \frac{L^2}{\tau}$   $T_{rot} = \frac{1}{2} \mathcal{I} \omega^2$ 

 $\mathcal{I}$  is the moment of inertia

E = T + V is negativ .... T is kinetic energy



$$\begin{split} &I_n \text{ QM it looks different} \\ &3\text{-dim Schr.Eq.} \to \text{Seperation of variables} \\ &x, y, z, \to r, \nu, \varphi \\ &\frac{\delta^2}{\delta x^2} + \frac{\delta^2}{\delta y^2} + \frac{\delta^2}{\delta z^2} \longrightarrow T_r + \frac{L^2(\theta, \varphi)}{r^2} \\ &T_r \longrightarrow \frac{1}{r^2} \frac{\partial}{\partial r} r^2 \frac{\partial}{\partial r} \end{split}$$

 $L^2$  is ugly (but can be made very elegant) This is generally used in many fields.

$$\frac{L^2(\theta,\varphi)}{r^2} \longrightarrow \frac{1}{r^2} \frac{1}{\sin\theta} \frac{\partial}{\partial \theta} \left( \sin\theta \frac{\partial \psi}{\partial \theta} \right) + \frac{1}{r^2 \sin^2\theta} \frac{\partial^2 \psi}{\partial \phi^2}$$

Solutions of the above are  $Y_l^m(\theta, \phi)$ , known as the spherical harmonics. (The constants were not taken care of)

From the web ("stolen latex code")

http://vergil.chemistry.gatech.edu/notes/quantrev/node25.html

$$-\frac{\hbar^2}{2\mu} \left[ \frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial \psi}{\partial r} \right) \frac{1}{r^2 \sin\theta} \frac{\partial}{\partial \theta} \left( \sin\theta \frac{\partial \psi}{\partial \theta} \right) + \frac{1}{r^2 \sin^2\theta} \frac{\partial^2 \psi}{\partial \phi^2} \right] - \frac{e^2}{4\pi\epsilon_0 r} \psi(r,\theta,\phi) = E\psi(r,\theta,\phi)$$
(1)

Exercise: Look on the separation of variables and how it's done.

(l=0) s-states, (l=1) p-states,

(l=2) d-states,

(l=3) f-states, ... there are more, but it is not relevant in typical atoms