## This note was written in "realtime" by Ingjald Pilskog. Edited by L. K. Third lecture 29.08.06

Atomic world construction  $a_0$  - radius, length  $\Delta x \approx a_0 \Delta k \approx 1/a_0$  $(\hbar \Delta k = \Delta p$  wavenumber  $\Delta E \approx \frac{(\Delta p)^2}{2m} = \frac{\hbar^2}{2m\epsilon}$  $\frac{\hbar^2}{2ma_0^2}\approx \frac{\hbar^2}{ma}$  $\frac{\hbar^2}{m a_0^2}$  (kinetic energy T<sub>0</sub>. ) Potential energy V<sub>0</sub>  $|-\frac{e^2}{a_0}$  $a_0$ Atomic unit of length  $a_0 = 0,529 \text{ Å} = \frac{\hbar^2}{e^2\pi}$  $\frac{\hbar^2}{e^2m}$  Lenght unit Atomic unit of energy Energy unit  $\frac{e^2}{a}$  $\frac{e^{\check{2}}}{a_0} = \frac{\check{\hbar}}{m c}$  $\frac{\hbar}{ma_0^2}=27,2\,\,{\rm eV}$  $\langle T_0 \rangle = \langle V_0 \rangle$  should be of the same order. Atomic unit of velocity  $p_0 = \hbar k_0 = \frac{\hbar}{ac}$  $\frac{\hbar}{a_0}$   $v_0 = \frac{p_0}{m} = \frac{\hbar}{m}$ m 1  $\frac{1}{a_0} = \frac{me^2}{\hbar^2}$  $\frac{\hbar}{m}=\frac{e^2}{\hbar}$  $\frac{v_0}{\hbar} = \frac{e^2}{\hbar c} = \alpha = \frac{1}{13}$  $137$ <br>Atomic unit of time<br>Atomic unit of time  $t_0 = \frac{a_0}{v_0}$  $\frac{a_0}{v_0} = (\frac{a_0}{e^2})\hbar = \frac{\hbar}{E_0}$  $E_0$ Alternative postulate  $t_0 = \frac{\hbar}{E}$ (Statement a.u.  $\longleftrightarrow e = m\hbar = 1$  is useless)  $\hbar = 0, 66 \cdot 10^{-15}$  eVs  $t_0 =$  $0,66·10<sup>-15</sup>$  $\frac{3 \cdot 10^{-15}}{27,2}$ s=0, 24 · 10<sup>−15</sup>  $(2 \pi$  for angular freq.)  $\nu$  frequency  $\omega$  - angular frequency  $k_0T$  is the "physical temperature". Room temperature is thus  $\frac{1}{40}$ eV or 25 meV Atomic unit of energy  $\longrightarrow$  VERY HOT

## More about bound states in H



States characterized by  $n, l$  ( $m$  - magnetic)

In H energies given by  $\frac{1}{n^2}(-\frac{1}{2})$  $(\frac{1}{2})$ a.u.

"Ladi says its nonsens to talk about m as a quantum number", wrote Ingjald. He did not write: "unless we have magnetic field on".

Angular momentum

$$
L = \omega \mathcal{I} \qquad T_{rot} = \frac{1}{2} \frac{L^2}{\mathcal{I}} \qquad T_{rot} = \frac{1}{2} \mathcal{I} \omega^2
$$

 $I$  is the moment of inertia

 $E = T + V$  is negativ .... T is kinetic energy



 $I_n$  QM it looks different 3-dim Schr.Eq.  $\rightarrow$  Seperation of variables  $x, y, z, \rightarrow r, \nu, \varphi$  $\frac{\delta^2}{\delta x^2} + \frac{\delta^2}{\delta y^2} + \frac{\delta^2}{\delta z^2} \longrightarrow T_r + \frac{L^2(\theta,\varphi)}{r^2}$  $r^2$  $T_r \longrightarrow \frac{1}{r^2} \frac{\partial}{\partial r} r^2 \frac{\partial}{\partial r}$ ∂r

 $L^2$  is ugly (but can be made very elegant) This is generally used in many fields.

$$
\frac{L^2(\theta,\varphi)}{r^2} \longrightarrow \frac{1}{r^2} \frac{1}{\sin\theta} \frac{\partial}{\partial\theta} \left( \sin\theta \frac{\partial\psi}{\partial\theta} \right) + \frac{1}{r^2 \sin^2\theta} \frac{\partial^2\psi}{\partial\phi^2}
$$

Solutions of the above are  $Y_l^m(\theta, \phi)$ , known as the spherical harmonics. (The constants were not taken care of)

From the web ("stolen latex code") http://vergil.chemistry.gatech.edu/notes/quantrev/node25.html

$$
-\frac{\hbar^2}{2\mu} \left[ \frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial \psi}{\partial r} \right) \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left( \sin \theta \frac{\partial \psi}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 \psi}{\partial \phi^2} \right] - \frac{e^2}{4\pi \epsilon_0 r} \psi(r, \theta, \phi) = E \psi(r, \theta, \phi)
$$
\n(1)

Exercise: Look on the separation of variables and how it's done.

 $(l=0)$  s-states,

 $(l=1)$  p-states,

 $(l=2)$  d-states,

 $(l=3)$  f-states, ... there are more, but it is not relevant in typical atoms